


Methods of Large-Scale Capture and Removal of Atmospheric Greenhouse Gases

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1. Introduction

The terms “global warming” and “climate change” refer to the large-scale impacts of human actions such as the burning of fossil fuels and extensive deforestation that contribute to a rise in the level of greenhouse gases in the atmosphere [1]. The Intergovernmental Panel on Climate Change (IPCC) asserts that the problem of greatest concern for humankind in this century is climate change, stimulated by an increase in the global average surface temperature of 1.1 °C since 1900 [2,3]. The global water cycle is being disrupted by climate change, resulting in droughts in some regions and extreme rainfall and flooding in others [4]. Continued warming will accelerate permafrost melting, which can cause a loss of seasonal snow cover, melting glaciers, and a reduction in the summer Arctic sea ice. Coastal communities across the world will be impacted, and lowland areas will see more frequent floods and severe shoreline erosion as sea levels rise [5]. More frequent ocean heat waves, ocean acidification, and reduced oxygen levels in the sea are all consequences of climate change that can have an impact on marine ecosystems and the livelihoods of people who depend on them. Cities, in particular, suffer significant effects of climate change. Urbanization increases the severity and frequency of heat waves as well as extreme precipitation [6]. The “Paris Agreement”, which seeks to restrict the rise in the average global temperature above its preindustrial level to 1.5 to 2 °C, was signed by 175 nations with the goal of minimizing the harm that is caused by climate change [7]. Removing greenhouse gases (GHGs) from the atmosphere is one important way to help achieve this target.

2. Greenhouse Gases in the Atmosphere

Carbon dioxide (CO₂), methane (CH₄), nitrous oxide (N₂O), hydrofluorocarbons (HFCs), perfluorocarbons (PFCs), and sulfur hexafluoride (SF₆) are the six GHGs that must be restricted in the atmosphere, according to the Kyoto Protocol [8]. The three most significant GHGs, with their respective average global concentration and their contribution to global warming, are: CO₂, 410.5 ppm, 66%; CH₄, 1877 ppb, 16%; N₂O, 332.0 ppb, 7%. The leading contributor to global warming is CO₂. The increase in average global temperature will reach 3.8 °C if the concentration of CO₂ in the atmosphere doubles, and it will drop by 3.6 °C if it is cut in half [9]. CH₄ presently contributes less to global warming, but its effect cannot be ignored. One ton of CH₄ causes as much global warming as 86 tons of CO₂ does [10]. Radiative forcing could be decreased by 16% within 10–20 years by returning the atmospheric CH₄ concentrations to the pre-industrial levels of 760 ppb [11]. It is therefore desirable—even imperative—to hold the amounts of CO₂ and CH₄ in the atmosphere to specific levels.



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2.1. Capture and Removal of CO₂

Strategies such as afforestation and vegetation restoration are natural and eco-friendly approaches to reducing the level of CO₂ in the atmosphere. The potential of nature to absorb CO₂, however, will take more than tens of thousands of years to achieve the goal of carbon neutrality if we are reliant just on the ecosystem. It is not practical to address climate change only by planting a great many trees quickly. Other approaches are also needed. Carbon capture and storage (CCS) is a way to reduce carbon emissions by capturing and storing CO₂ at its source [12].

CO₂ can be captured in a variety of ways, including electrochemically [13], through liquid-absorbency [14], and physically [15,16]. The cost of capturing one ton of CO₂ with CCS technology is \$36 to \$53 [17], which is cost-effective. The active deployment of CCS technology has not been widely adopted because there are still some technical challenges to be solved [18,19]. Although CCS can dramatically lower emissions from the electric-generation sector (coal-fired power plants), it is unable to capture emissions from the construction and transportation sectors, making it challenging for CCS technology alone to reach the goal of the Paris Agreements. Active emission reduction strategies can more effectively address dispersed carbon emission sources and accomplish decarbonization. The most feasible alternate technique, known as bio-energy with carbon capture and storage (BECCS), involves absorbing CO₂ from the atmosphere by having plants or crops grow and with the use of CCS technology [20]. The trees may be burnt for energy while the CO₂ that is emitted during the combustion is captured using the CCS technology. The trapped CO₂ is kept underground, where it is preserved from escaping back into the atmosphere, and the process is repeated. The technique has the potential to eventually take all of the excess CO₂ out of the atmosphere if it is used on a large enough scale [21]. However, BECCS requires a large forested area, which can result in a shortage of farm land and fresh water, and have other negative consequences [22]. Direct air capture (DAC) is the strategy of chemically directly converting CO₂ from the atmosphere into compounds like carbonates [23]. Figure 1 shows that relative to BECCS, DAC can lessen the use of land and water, but its high cost is still a major barrier to its commercialization. The standard DAC system costs around \$600 per ton of CO₂ that is captured [24]. According to a recent report [25], a pilot plant can decrease the cost to \$113–232 per ton of CO₂ that is captured by implementing appropriate energy-saving measures, but this is still significantly more expensive than CCS.

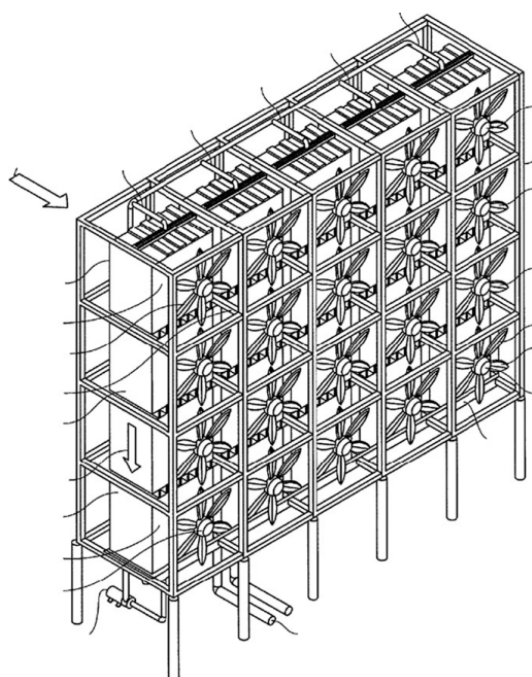


Figure 1. Schematic of the DAC device [26].

2.2. Capture and Removal of CH₄

In fact, planning only for the removal of CO₂ from the atmosphere is not enough. Even if no additional CO₂ is produced, the existing carbon store (515 Gt) can continue to contribute to global warming for decades to come [27]. The oceans also store a substantial amount of CO₂. It is released into the atmosphere when the level of CO₂ in the atmosphere drops due to the breakdown of carbonic acid in the seas [28].

Reducing atmospheric CH₄ levels, according to the United Nations Environment Programme (UNEP), is another efficient strategy to combat climate change [29]. It can reduce warming by 0.4 to 0.5 °C by 2050. Rising ocean temperatures will cause CH₄ (1146 Gt) in the form of flammable ice to become unstable and release itself into the atmosphere, thereby exacerbating global warming [30]. There are few experiments currently being done on atmospheric CH₄ removal. Thermal catalysis is regarded by certain researchers as a potent technique for removing atmosphere CH₄. The disadvantages of thermal catalysis, which frequently involves high temperatures and pressures, include considerable energy consumption, difficult reaction conditions, and some safety risks. CH₄ has a high C-H bond energy (413 kJ/mol) and a relatively stable structure. It is presently not economical to employ thermal catalytic technology on a wide scale to degrade atmospheric CH₄ because it demands temperatures of more than 700 °C. The limit of the reaction conditions can be lowered by catalyst improvement. Brenneis et al. [31] added copper-treated zeolite particles to a reaction tube through which air passed. The zeolite was capable of capturing and converting all of the CH₄ in the air when heated it was to 310 °C. The only place it can actually be utilized is in a lab, and even there, the reaction temperature is too high for the process to be practical.

Photocatalytic semiconductor technology can be used to remove pollutants at room temperature and pressure, and the most common pollutants, such as indoor formaldehyde and volatile organic compounds (VOCs), have achieved good degradation rates. De Richter [32] presented a solar chimney power plant that was integrated with photocatalytic reactors (SCPP-PCRs) to address the problem of climate change. The system utilizes a natural convection mechanism to provide a powerful airflow and a driving force without the need for fuel. The canopy is coupled with a photocatalytic reactor. Atmospheric CH₄ that comes into contact with the photocatalyst is converted to CO₂ and H₂O under solar irradiation. Ming et al. [33] confirmed the feasibility of SCPP-PCRs in combating global climate change by using a numerical simulation method, as shown in Figure 2. A good photocatalyst is required for the system to achieve the best performance. Chen et al. [34] carried out experiments on the oxidation of CH₄ by a new synthetic photocatalyst (Ag/ZnO), and they found that nano-scale zinc oxide could effectively oxidize 100% of CH₄ under sunlight irradiation, and the activity of the silver-plated ZnO semiconductor remained stable after multiple uses. The reaction rate depended on the concentration of CH₄; it was faster at lower concentrations and less susceptible to temperature changes after the light conditions were set up. The experimental findings demonstrate that the photocatalytic technique that is utilizing Ag-ZnO has special benefits for low levels of atmospheric CH₄ that are challenging to handle by thermal catalysis. Other photocatalysts with great potential include g-C₃N₄@Cs_{0.33}WO₃, ZnO/CuO, and Ga₂O₃/AC. The DAC system can still be used to capture and degrade atmospheric CH₄ by creating a continuous stream of air with a fan, then absorbing the CH₄ with organic solutions or removing it using a honeycomb photocatalytic reactor that is positioned behind the fan [35]. However, if the chemical absorption is considerable, then the employment of the DAC results in excessive energy consumption and subsequent maintenance expenses.

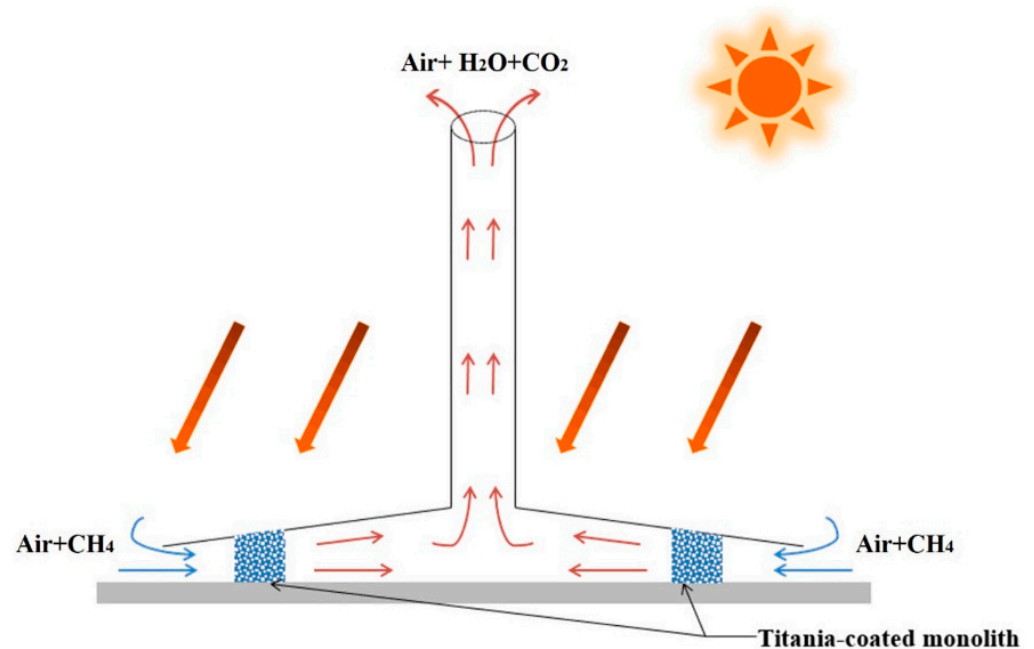


Figure 2. Schematic of the SCPP-PCRs [33].

3. Conclusions

It is essential to continue studying methods for removing greenhouse gases from the atmosphere. The following are a few major conclusions:

1. The disadvantages of the current methods for quick, large-scale CO₂ removal from the atmosphere can include significant energy consumption, high investment, and high maintenance costs. Transformative technologies are needed in this field.
2. The removal of atmospheric CH₄ is a worthwhile goal, but because this gas is so rarefied, and removing it remains a challenge.
3. Although the study of SCPP-PCRs is still in its early stages, it appears to be a promising technology that can not only decrease CO₂ emissions from thermal power plants but also the diminish greenhouse gases that are in the air.

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